

Directions for use - I : (P. Alkalinity & Total Alkalinity)

1. Take 10 ml of water sample to be tested in the Test jar.
2. Add 2 drops of Reagent ALK-1. Mix contents well.
3. If pink colour appears, it indicates presence of P. Alkalinity. (If pink colour does not appear then P. Alkalinity is zero & hence follow Directions for Use-II given at the reverse).
4. Now add Reagent ALK-3 drop wise, counting the number of drops while mixing until **the PINK colour disappears (say X drops)**.
5. To this solution, add 2 drops of Reagent ALK-2. The sample will turn Yellowish Orange.
6. Now drop wise add reagent ALK-3, counting the number of drops while mixing until **the colour changes from YELLOWISH ORANGE TO PINK ORANGE (say Y drops)**.

If the expected Alkalinity of the sample is more than 100 ppm then use Reagent ALK-4 instead of Reagent ALK-3.

Calculations :

P. Alkalinity ppm as CaCO ₃	= 5 x 'X' drops of Reagent ALK-3.
	= 50 x 'X' drops of Reagent ALK-4.
Total Alkalinity ppm as CaCO ₃	= 5 x (X+Y) drops of Reagent ALK-3.
	= 50 x (X+Y) drops of Reagent ALK-4.

Directions for use - II : (For Total Alkalinity only)

1. Take 10 ml of water sample to be tested in the Test jar.
2. Add 2 drops of Reagent ALK-2. The sample will turn **Yellowish Orange**
3. Now add reagent ALK-3 drop wise, counting the number of drops while mixing until **the colour changes from YELLOWISH ORANGE TO PINK ORANGE**.

If the expected Alkalinity of the sample is more than 100 ppm then use Reagent ALK-4 instead of Reagent ALK-3.

Calculations :

Total Alkalinity as ppm CaCO₃ = 5 x Number of drops of Reagent ALK-3
 = 50 x Number of drops of Reagent ALK-4

Calculation of Caustic (Hydroxide), Carbonate & Bicarbonate Alkalinity from Total Alkalinity (T) & Phenolphthalein Alkalinity (P)

Value of P & T	Caustic Alkalinity	Carbonate Alkalinity	Bicarbonate Alkalinity
P = 0	0	0	T
P < ½ T	0	2P	T – 2 P
P = ½ T	0	2P	0
P > ½ T	2P – T	2(T – P)	0
P = T	T	0	0

Alkalinity**Code : XL-113****Range : 10 - 200 & 100 – 2,000 ppm as CaCO₃****AQUA-XL**Water Analysing
Kits**Directions for use – I : (P. Alkalinity & Total Alkalinity)**

1. Take 10 ml of water sample to be tested in the Test jar.
2. Add 2 drops of Reagent ALK-1. Mix contents well.
3. If pink colour appears, it indicates presence of P. Alkalinity .(If pink colour does not appear then P. Alkalinity is zero & hence follow Directions for Use-II given at the reverse).
4. Now add Reagent ALK-5 drop wise, counting the number of drops while mixing until **the PINK colour disappears (say X drops)**.
5. To this solution, add 2 drops of Reagent ALK-2. The sample will turn Yellowish Orange
6. Now drop wise add reagent ALK-5, counting the number of drops while mixing until **the colour changes from YELLOWISH ORANGE TO PINK ORANGE (say Y drops)**.

If the expected Alkalinity of the sample is more than 200 ppm then use Reagent ALK-6 instead of Reagent ALK-5.

Calculations :

$$\begin{aligned} \text{P. Alkalinity ppm as CaCO}_3 &= 10 \times \text{'X'} \text{ drops of Reagent ALK-5.} \\ &= 100 \times \text{'X'} \text{ drops of Reagent ALK-6.} \\ \text{Total Alkalinity ppm as CaCO}_3 &= 10 \times (\text{X}+\text{Y}) \text{ drops of Reagent ALK-5.} \\ &= 100 \times (\text{X}+\text{Y})\text{drops of Reagent ALK-6.} \end{aligned}$$

Alkalinity**Code : XL-113****Range : 10 - 200 & 100 – 2,000 ppm as CaCO₃****AQUA-XL**

Water Analysing Kits

Directions for use - II : (For Total Alkalinity only)

1. Take 10 ml of water sample to be tested in the Test jar.
2. Add 2 drops of Reagent ALK-2. The sample will turn **Yellowish Orange**.
3. Now add reagent ALK-5 drop wise, counting the number of drops while mixing until **the colour changes from YELLOWISH ORANGE TO PINK ORANGE**.

If the expected Alkalinity of the sample is more than 200 ppm then use Reagent ALK-6 instead of Reagent ALK-5.

Calculations :

$$\begin{aligned} \text{Total Alkalinity as ppm as CaCO}_3 &= 10 \times \text{Number of drops of Reagent ALK-5} \\ &= 100 \times \text{Number of drops of Reagent ALK-6} \end{aligned}$$

Calculation of Caustic (Hydroxide), Carbonate & Bicarbonate Alkalinity from Total Alkalinity (T) & Phenolphthalein Alkalinity (P)

Value of P & T	Caustic Alkalinity	Carbonate Alkalinity	Bicarbonate Alkalinity
P = 0	0	0	T
P < ½ T	0	2P	T – 2 P
P = ½ T	0	2P	0
P > ½ T	2P – T	2(T – P)	0
P = T	T	0	0